

# Fluid Equations.



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- Mechanics.
- Thermodynamics.
- Electromagnetism.

- Nature is described by a finite number of equations.
- The difference between reality and the computed results:
  - Some minor physical effects are neglected.
  - Numerical approximations.(Iterative processes, Roundoff errors, etc...)
  - Geometrical approximations.

- Newton's Laws

- First law: The velocity of a body remains constant unless the body is acted upon by an external force.
- Second law: The variation of a body momentum in time is parallel and directly proportional to the net force  $\mathbf{F}$  and inversely proportional to the mass  $m$ .

$$\mathbf{F} = \frac{d\mathbf{p}}{dt}$$

- Third law: The mutual forces of action and reaction between two bodies are equal, opposite and collinear.

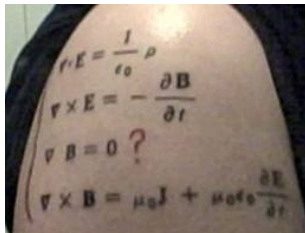
And God said...

$$\nabla \times \vec{E} = -\mu \frac{\partial \vec{H}}{\partial t}$$

$$\nabla \times \vec{H} = \vec{J} + \varepsilon \frac{\partial \vec{E}}{\partial t}$$

$$\nabla \cdot \vec{D} = \rho$$

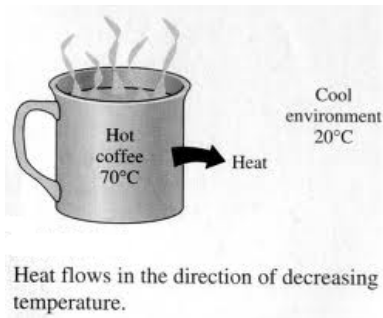
$$\nabla \cdot \vec{B} = 0$$



and there was light.

# Thermodynamics

- First law of thermodynamics, about the conservation of energy:
  - Energy is neither created nor destroyed.
  - There is no free lunch.
- Second law of thermodynamics, about entropy:
  - In an isolated system, the entropy never decreases.
  - Heat cannot spontaneously flow from a colder location to a hotter area – work is required to achieve this.



- Particular case of classical Mechanics.
- Core ideas coming from Thermodynamics.
- Extension of Newton's law to a complex system.



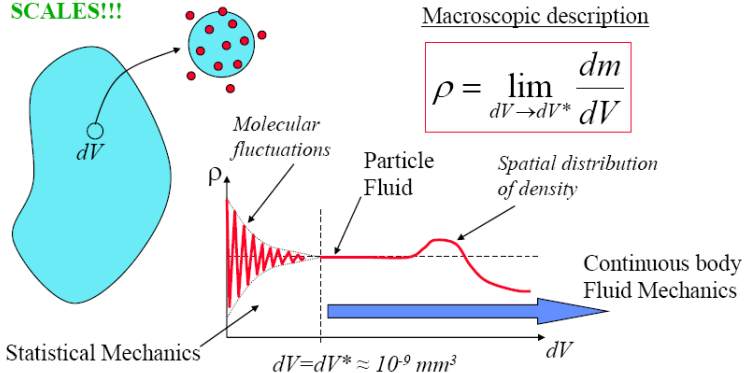
- Continuum hypothesis.
- Mass conservation.
- Second Newton's law.  $\mathbf{F} = \frac{d\mathbf{p}}{dt}$
- First Principle of Thermodynamics: Energy conservation.



# Continuum hypothesis.

- Fluids have a molecular nature.  $V = 10^{-9} \text{ mm}^3$  air in NC  
 $\Rightarrow 3 \cdot 10^7$  molecules

**SCALES!!!**



- Mathematical concept.
- What size?
  - Large enough to contain many molecules.
  - Small enough to allow the use of the differential calculus.
- Hypothesis: Local thermodynamic equilibrium.
  - Random free walk  $\lambda \ll$  Problem dimensions.
  - Average time between molecular collisions  $\ll$  rate of change of the fluid variables.
- Conclusion: Finite volume (fluid particle) is defined by one velocity  $\mathbf{v}$ , pressure  $p$ , density  $\rho$ , Temperature  $T$ , etc...

# How to work with fluids.

- Fundamental laws:
  - Mass conservation.
  - Momentum conservation.
  - Energy conservation.
- Extra information:
  - Equations of state.
  - Boundary conditions.

⇒ System of non-linear partial differential equations.

- Difficult analytical solution
- Expensive and difficult experiments.

⇒ **Numerical solution.**

# Fluid equations for a fluid volume.

- $M$  total mass of our fluid volume.
- $\mathbf{P}$  total momentum of our fluid volume.
- $\mathbf{F}$  total force that our fluid is experiencing.
- $E$  total energy that our fluid contains.
- $Q$  total heat that our fluid is transferring.
- $W$  total work that our fluid is performing.

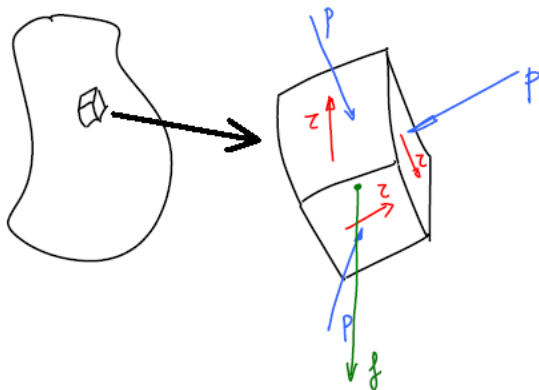
$$\frac{dM}{dt} = 0$$

$$\frac{d\mathbf{P}}{dt} = \mathbf{F}$$

$$\frac{dE}{dt} = Q + W$$

# Forces over a fluid particle.

- External forces  $f$ . Examples: gravity, electromagnetic, inertial, etc...
- Friction forces. The particles experience the force by physical contact. Examples: **viscous friction** and **pressure**.



# Fluid equations of a fluid particle: Navier-Stokes equations.

And God said again:

- Incompressible fluid.
- Newtonian fluid.
- Only mechanical properties will be considered, thermal effects will be neglected.

$$\begin{aligned}\nabla \cdot \mathbf{v} &= 0 \\ \rho \frac{\partial \mathbf{v}}{\partial t} + \rho \mathbf{v} \cdot \nabla \mathbf{v} &= -\nabla p + \rho \mathbf{f} + \nabla \tau\end{aligned}$$

- $\mathbf{f}$  external volumetric forces: gravity, electromagnetic, inertial, etc...
- $\tau$  viscous stress tensor.

- Compressibility: possibility of density changes.
  - Liquids always incompressible  $\rho = \text{const.}$
  - Gases are compressible under some hypothesis.
- Laminar and turbulent.
- Stationary and time dependant.
- Unidirectional: One velocity component.

Navier-Stokes Equations. Incompressible without thermal effects  $\mu = \text{cons.}$

$$\nabla \cdot \mathbf{v} = 0$$

$$\rho \frac{\partial \mathbf{v}}{\partial t} + \rho \mathbf{v} \cdot \nabla \mathbf{v} = -\nabla p + \rho \mathbf{f} + \mu \nabla^2 \mathbf{v}$$